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of the reactions will occur. For those that do not, write no reaction. I.

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the oxidation number

of the underlined

element: KMnO_4

Since K is an alkali

metal, its charge must

be $1+$ Oxygen is $2-$

but there are four of

them, therefore, 4

times $2-$ equals $8-$ If

$1+$ and $8-$ are added

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many of the

transuranium

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following names:

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apart and are in net
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