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CHAPTER 3 ANSWERS AND EXPLANATIONS. Multiple-Choice. 1. D This element has five peaks, meaning a total of five subshells. The final peak, which would be located in the 3p subshell, is slightly higher than the 3s peak to the left of it. A full 3s peak has two electrons, therefore there must be at least three electrons in the 3p subshell. The element that best fits this is phosphorus. 2.

CHAPTER 3 ANSWERS AND EXPLANATIONS - Atoms, Elements, and ...

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3 Atoms: The Building Blocks of Matter

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