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⌘ Polyatomic Ions - Chemistry

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~~Chemical compounds name and their  
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9.1. Naming & Writing Formulas for Ionic  
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Writing Formulas for Acids: Chapter 9.4.  
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## Chapter 9: Chemical Names & Formulas

oxygen b. iodine c. sodium e. nickel, 2  
electrons lost d. aluminum f. magnesium

2. How many electrons does the neutral  
atom gain or lose When each ion forms? a.  
b.  $\text{p}^{3-}$  d.  $\text{Ca}^{2+}$  / Z 3. Name each I each as a  
cation Or an anion. f.  $\text{O}^{2-}$  f.  $\text{Mn}^{2+}$  4. Write  
the formula (including charge) for each

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Mister Chemistry Welcomes You! ☐

Chemistry teacher at ...

The chemical formula of a compound is a symbolic representation of its chemical composition. Chemical formulae provide insight into the elements that constitute the

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Answers  
molecules of a compound and also the  
ratio in which the atoms of these elements  
combine to form such molecules.

Chemical Formula of Common  
Compounds, Chemical reaction ...  
ions). Chapter 9 □ Chemical Names and  
Formulas Chapter 9 Chemical Names and

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Formulas 223 2 Complete this table by writing correct formulas for the compounds formed by combining positive and negative ions 3 Name the following compounds a K<sub>3</sub>PO<sub>4</sub> c NaHSO<sub>4</sub> e N<sub>2</sub>O<sub>5</sub> g PI<sub>3</sub> b Al(OH)<sub>3</sub> d HgO f NBr Chapter 9 Chemical Names Formulas Worksheet

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Chapter 9 Chemical Names And Formulas  
Practice Problems ...

1. Chapter 9 Chemical Names and  
Formulas H<sub>2</sub>O 1. 2. Section 9.1 Naming



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**OBJECTIVES:** Identify the charges on monatomic ions by using the periodic table, and name the ions. Define a polyatomic ion and write the names and formulas of the most common polyatomic ions. Identify the two common endings for the names of most polyatomic ions.

**Atoms and Ions** Atoms are

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## Answers

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Formulas - Notes

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Chapter 9: Chemical Names and  
Formulas. Honours Chemistry D-  
period Ms. Steele. STUDY. PLAY.

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Monatomic ion. - single atom with a positive or negative charge resulting from the loss or gain of electrons, respectively.

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## Chapter 9 Chemical Names And Formulas

chapter 9 : chemical names and formulas.

balancing the formulas. stock system.

classical system. oxidation rules. 1. write

the element with the lower

electronegativity first (g. - the oxidation

number (charge) is written in parenthesis

beh. - in naming use and -ous for the

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chapter 9 chemical names and formulas

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Complete the following five rules for writing a chemical formula from a chemical name. a. In an ionic compound, the net ionic charge is \_\_\_\_\_. b. An -ide ending generally indicates a \_\_\_\_\_

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Answers  
compound. c. An -ite or -ate ending means  
there is a \_\_\_\_\_ ion that includes oxygen  
in the formula. CHAPTER 9, Chemical

Name Date Class CHEMICAL NAMES  
AND FORMULAS 9

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Monatomic ions. Polyatomic ions. Binary  
compound. Acid. A single atom with a  
positive or negative charge. A tightly  
bound group of atoms that behaves as a  
unit ...

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Chapter 9 Chemical Names And Formulas  
Practice Problems ...

Which set of chemical name and chemical formula for the same compound is correct

- A. Ammonium sulfite,  $(\text{NH}_4)_2\text{S}$  B.  
Iron(III)phosphate,  $\text{FePO}_4$  C. Lithium  
carbonate,  $\text{LiCO}_3$

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Names and Formulas. STUDY. PLAY.  
Monatomic ions. Consist of a single atom  
with a positive or negative charge  
resulting from the loss or gain of one or

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more valence electrons, respectively. Key

## 1 Answers

Chapter 9 Chemical Names And Formulas  
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Chapter 9: Chemical Names and Formulas  
Section 9., Naming Ions The language of  
Chemistry.. Chemistry has its own



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Answers

distinct language □ With chemical names and formulas, it can all seem a bit daunting □ It is important for you to understand how Ions and ionic compounds are named so you can understand many aspects of this □language□

Chapter 9: Chemical Names and Formulas

*Page 33/36*

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